

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713597286>

Nucleosides, III¹ Synthesis and Properties of 2-Trifluoromethyl-Naphthimidazole-Ribonucleoside

Abdullah Hijazi^a

^a Dept. of Chemistry, King Saud University, Riyadh, Saudi Arabia

To cite this Article Hijazi, Abdullah(1986) 'Nucleosides, III¹ Synthesis and Properties of 2-Trifluoromethyl-Naphthimidazole-Ribonucleoside', *Nucleosides, Nucleotides and Nucleic Acids*, 5: 5, 529 — 537

To link to this Article: DOI: 10.1080/07328318608068695

URL: <http://dx.doi.org/10.1080/07328318608068695>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

NUCLEOSIDES, III¹
SYNTHESIS AND PROPERTIES OF 2-TRIFLUOROMETHYL-NAPHTHIMIDAZOLE-
RIBONUCLEOSIDE

Abdullah Hijazi*

* Dept. of Chemistry, King Saud University,
Riyadh, Saudi Arabia.

Abstract. The fusion reaction between 2-trifluoromethyl-naphth[2,3-d]imidazole (1) and 1-O-acetyl-2,3,5-tri-O-benzoyl-D-ribofuranose (2) leads to 2',3',5'-tri-O-benzoyl-1- β -D-ribofuranosylnaphth[2,3-d]imidazole (3). Debenzoylation of (3) gives the corresponding nucleoside 1- β -D-ribofuranosyl-2-trifluoromethylnaphth[2,3-d]imidazole (4). Structural proofs are based on elementary analysis, UV-and ¹H-NMR spectra.

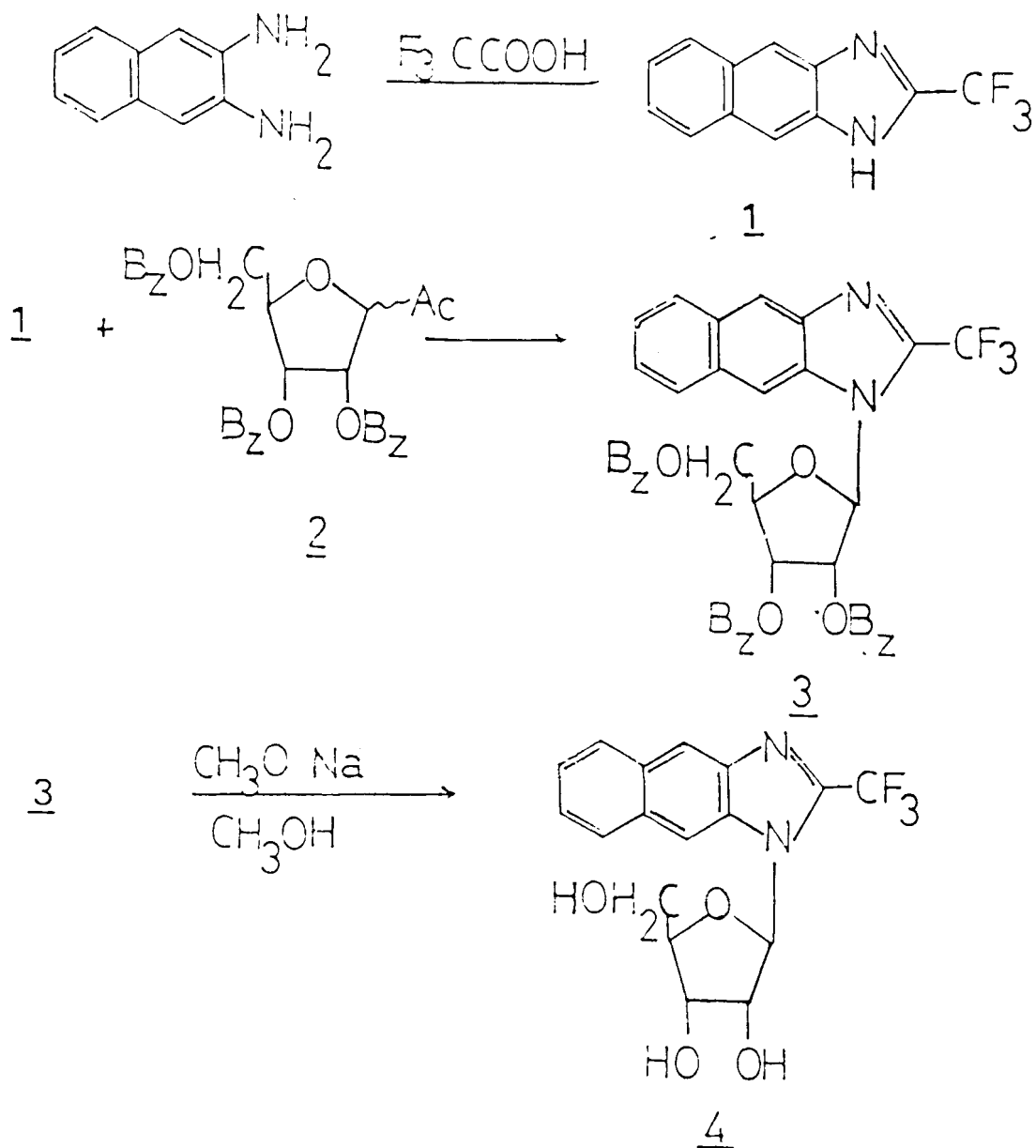
Considerable interest has been evinced in recent years in studying the effect of altering the heterocyclic moiety of a biologically active nucleoside. Antibacterial² as well as antiviral³ properties of some 2-trifluoromethylbenzimidazole-derivatives have been described. These derivatives are

known as inhibitors of photosynthetic processes⁴; some of them also exhibit appreciable herbicidal and insecticidal activities⁵. It was, therefore, of interest to study the ribosylation of the 2-trifluoromethyl-lin-naphth[2,3-d]-imidazole system. The present study is, in fact, an extension of our former investigations with lin-naphth[2,3-d]imidazole⁶.

Such compounds have a potential biological significance, due to the fact that the lin-naphthimidazole-cobalamine analog⁷ has been isolated as a minor vitamin B₁₂ component. The starting material 2-trifluoromethylnaphth[2,3-d]imidazole (1) was obtained from 2,3-diaminonaphthalene and trifluoroacetic acid by a known procedure⁸.

The first attempt to achieve ribosylation of (1), using the modified fusion method, developed in the benzimidazole series⁹, was a failure and the expected nucleoside was not formed. A second attempt carried out using trimethylsilyl triflate, a new catalyst described by Vorbruggen et al¹⁰, also failed.

After the failing of the two previous attempts, we decided to examine the fusion of (1) with (2) at 180°C under reduced pressure. This proved to be successful and gave 1-(2,3,5-tri-O-benzoyl)- β -D-ribofuranosyl-2-(trifluoromethyl)-naphth[2,3-d]imidazole (3) in 63% yield. Debenzoylation to the free riboside (4) was performed by Zemplen's method¹¹, using the pure anomer.



The empirical formulas of the newly synthesized compounds were determined by elementary analyses and the structural assignments could be derived unambiguously from UV and ^1H -NMR spectra.

Table 1. UV-Absorption Spectra of 2-Trifluoromethylnaphthimidazole Derivatives in MeOH.

2-Trifluoromethylnaphth [2,3-d]-imidazole	λ_{\max} (nm)	log ϵ
unsubstituted (1)	[302] 317 335 349	[3.63] 3.86 3.71 3.5
1-(2,3,5-Tri-O-benzoyl β -D-ribofuranosyl)-(3)	221 236 274 281 [306] 316 338, [352]	[4.77] 4.87 3.60 3.60 [3.78] 3.95 3.71 [3.59]
1- β -D-Ribofuranosyl(4)	219 239, [286][306]317 337.5 [352]	4.61 4.77 [3.40] [3.80] 3.96 3.78 [3.63]

Table 2. ^1H -NMR Spectra of Naphth[2,3-d]imidazole Derivatives in $\text{D}_6\text{-DMSO}$ and CDCl_3 *

	Aromatic Protons	1'-H(1) $J_{1',2'}$ (Hz)	2'-H 3'-H 4'-H 5'-H (1) (1) (1) (2)	2'-OH 3'-OH 5'-OH (1) (1) (1)
<u>1</u>	8.3bs(2) 8.03 dd(2) 7.49 dd(2)			
* <u>3</u>	7.30-8.12 m (21)	6.6 d 7.94	6.3pt 6.15dd 4.8dd 4.97m	
<u>4</u>	8.83s(1) 8.49(1) 8.08dd(2) 7.50m(2)	5.91d 7.63	4.72q 4.26m 4.07d 3.83m	5.47pt 5.47pt 5.29d

S = singlet; bs = broad singlet; d = doublet; dd = double doublet; pt = pseudotriplet;
q = quadruplet; m = multiplet. () number of protons.

The configuration of the glycosidic linkages can be assigned readily from the ^1H -NMR spectra taken in CDCl_3 and D_6 -DMSO, respectively (TAB. 2). As expected¹², the direct fusion reaction method resulted exclusively in the formation of β -anomer as indicated by an upfield chemical shift of $1'\text{-H}$. However, this is different from the chemical shift observed in the lower field which was recognized for α -anomer of similar system⁶. Again this result is in agreement with earlier report for β -anomer on very similar system¹², and other ribofuranosides¹³⁻¹⁷, that in an anomeric pair the chemical shift of the anomeric proton $1'\text{-H}$ of the α -D-riboside appears at lower field compared to that of the corresponding β -form. It is furthermore noticed that there is in almost all cases a very distinct separation and coupling of the sugar protons proving the assigned configuration conclusively. A comprehensive study is currently underway to determine the biological characteristics of the newly synthesized compound. Initial results from this study are very encouraging and a separate report is intended upon its completion.

Experimental

UV spectra were recorded on a Carey Recording Spectrophotometer, Model 118, from Appl. Physics Corp. ^1H -NMR spectra were obtained from Bruker WM 250. Thin layer chromatography was performed on silica gel sheets F1550 LS

254 of Schleicher & Schull, preparative thick layer chromatography on glass plates 40 x 20 cm coated with a 0.2 cm layer of silica gel PF₂₅₄ of Merck/Darmstadt and colum chromatography on Merck silica gel 60 (particle size 0.063-0.2 mm). Drying of the substances was achieved in a vacuum desiccator or in a Buchi-TO₅₀ drying oven under vacuum at room temp. and slightly elevated temp. respectively. Melting points are determined in a Tottoli apparatus and are uncorrected.

2-Trifluoromethylnaphth[2,3-d]imidazole (1). 2,3-Diaminonaphthalene (1.58 g, 0.01 mol) was refluxed with 10 ml of trifluoroacetic acid, resulting in the formation of a brownish precipitate. This precipitate was filtered and washed thoroughly with water before it was suspended again in water, and heated until boiling. Ethanol was then added gradually until the suspension was converted to a clear solution. Active charcoal was then added to the boiling solution and the entire contents were kept at boiling temp. for 10 min. The solution was then filtered and cooled to give colourless crystals (2.1 g, 89%) of m.p. 270°C. Lit. m.p. 273°C.

Anal. Calc. for C₁₂H₇N₂F₃ (236.2): C₁ 61.2; H, 2.99; N, 11.86 Found: C, 60.67; H, 2.93; N, 11.81.

1-(2,3,5-Tri-O-benzoyl- β -D-)-2-trifluoromethylnaphth
[2,3-d]imidazole (3).

A finely ground mixture (0.95 g 0.004 mol) of 2-trifluoromethylnaphth[2,3-d]imidazole and (2.4 g, 0.0047 mol) of 1-O-acetyl-2,3,5-tri-O-benzoyl- β -D-ribofuranose (2) was heated for two hours at 180–185⁰ under reduced pressure (20 mm Hg, water pump). The mixture was then taken in 50 ml methanol and refluxed for $\frac{1}{2}$ hr and filtered. The separation into the pure component (3) was achieved by preparative silica gel chromatography of 0.13 g from residue on a 40 x 20 x 0.2 cm plate and development in 1,2-dichloroethane/ethyl acetate (25/1) three times. The lower part was cut out and eluted by chloroform to yield, on evaporation, a colourless foam which was crystallized from ethanol to give the pure β -anomer (0.082 g, 63%) of m.p. 211⁰C.

Anal. Calc. for C₃₈H₂₇N₂O₇F₃ (680.59): C 67.05; H, 3.99; N, 4.11 Found C, 66.86; H, 3.81; N, 4.07.

1- β -D-Ribofuranosyl-2-trifluoromethylnaphth[2,3-d]
imidazole (4). Compound (3) (0.082 g, 0.00012 mol) was added to methanolic sodium methoxide solution (40 mg sodium in 100 ml of absol. methanol) and then stirred for 4 h at room temp. After addition of water (15 ml) the solution was neutralized with acetic acid and evaporated to dryness. The residue was coevaporated three times with water (10 ml), twice with methanol (20 ml) and then was crystallized

from water (10 ml) to give a colourless product (0.038g, 84%) of m.p. 188-9⁰.

Anal. Calc. for $C_{17}H_{15}N_2F_3O_4$ (370.41): C, 55.12; H, 4.08 N, 7.59. Found: C, 55.38; H, 4.16; N, 7.28.

ACKNOWLEDGEMENT

I thank Prof. Dr.Dr. W. Pfeleiderer and his laboratory staff for the work conducted in his lab to measure the UV and NMR spectra as well as the elementary analysis.

This research (Chem/1402/37) was supported by the Research Center, College of Science, King Saud University, Riyadh, Saudi Arabia.

REFERENCES

1. A. Hijazi and W. Pfeleiderer Nucleosides & Nucteotides (1986) (Part II: submitted).
2. B.C. Bishop, E.T.J. Chelton and A.S. Jones, Biochem. Pharmacol., 13, 751 (1964).
3. G.M. Fara and K.W. Cochran, Boll. dell' 1st. Sieroterapic Milan., 42, 630 (1964).
4. K.H. Buchel, F. Korte and R.B. Beechey, Angew. Chem. Internat. ED. 4 788 (1965).
5. D.E. Burton, A.J. Lambie, J.C.L. Ludgate, G.T. Newbold, A. Percival, and D.T. Sagggers, Nature (London) 208, 1166 (1965).

6. A. Hijazi and W. Pfeleiderer, Nucleosides & Nucleotides 3, 549, (1984).
7. W. Friedrich and K. Bernhauer, Angew. Chem. 71, 284 (1959).
8. K. Fries, R. Walter and K. Schilling, Liebigs Ann. Chem. 516, 248, (1935).
9. G.R. Revankar and L.B. Townsend, J. Heterocycl. Chem. 5, 477 (1968).
10. H. Vorbruggen, K. Krolikiewicz and B. Bennua, Chem. Ber. 114, 1234 (1981).
11. G. Zemplen, A. Geres and J. Hadacsy, Ber. Dtsch. Chem. Ges. 69, 1827 (1936).
12. Z. Kazimierczuk, R. Stolarski, L. Dudycz and D. Shugar, Nucleosides & Nucleotides, 1 (3), 275 (1982).
13. R.U. Lemieux, R.X. Kullnig, H.J. Bernstein and W.G. Schneider, J. Am. Chem. Soc. 80, 6098 (1958).
14. R.U. Lemieux and M. Hoffer, Can. J. Chem. 39, 110 (1961).
15. M.J. Robins and R.K. Robins, J. Am. Chem. Soc. 87, 4934 (1965).
16. T. Nishimura and B. Shimizu, Chem. Pharm. Bull. 13, 803, (1965).
17. I.W. Southon and W. Pfeleiderer, Chem. Ber. 111, 996 (1978).
18. K.H. Buchel, Z. Naturforsch. 25b, 934 (1970).

Received January 31, 1986.